

Agency for Innovation by  
Science & Technology

## *MatISSE/JP NM Workshop on Cross-Cutting Issues in Structural Materials R&D for Future Energy Systems*

# Exploring the Potential of **MAX Phases** for Select Applications in Extreme Environments

K. Lambrinou<sup>1</sup>, T. Lapauw<sup>1,2</sup>, J. Vleugels<sup>2</sup>

<sup>1</sup> SMR Group, NMS Institute, SCK•CEN; <sup>2</sup> Dept. MTM, KU Leuven

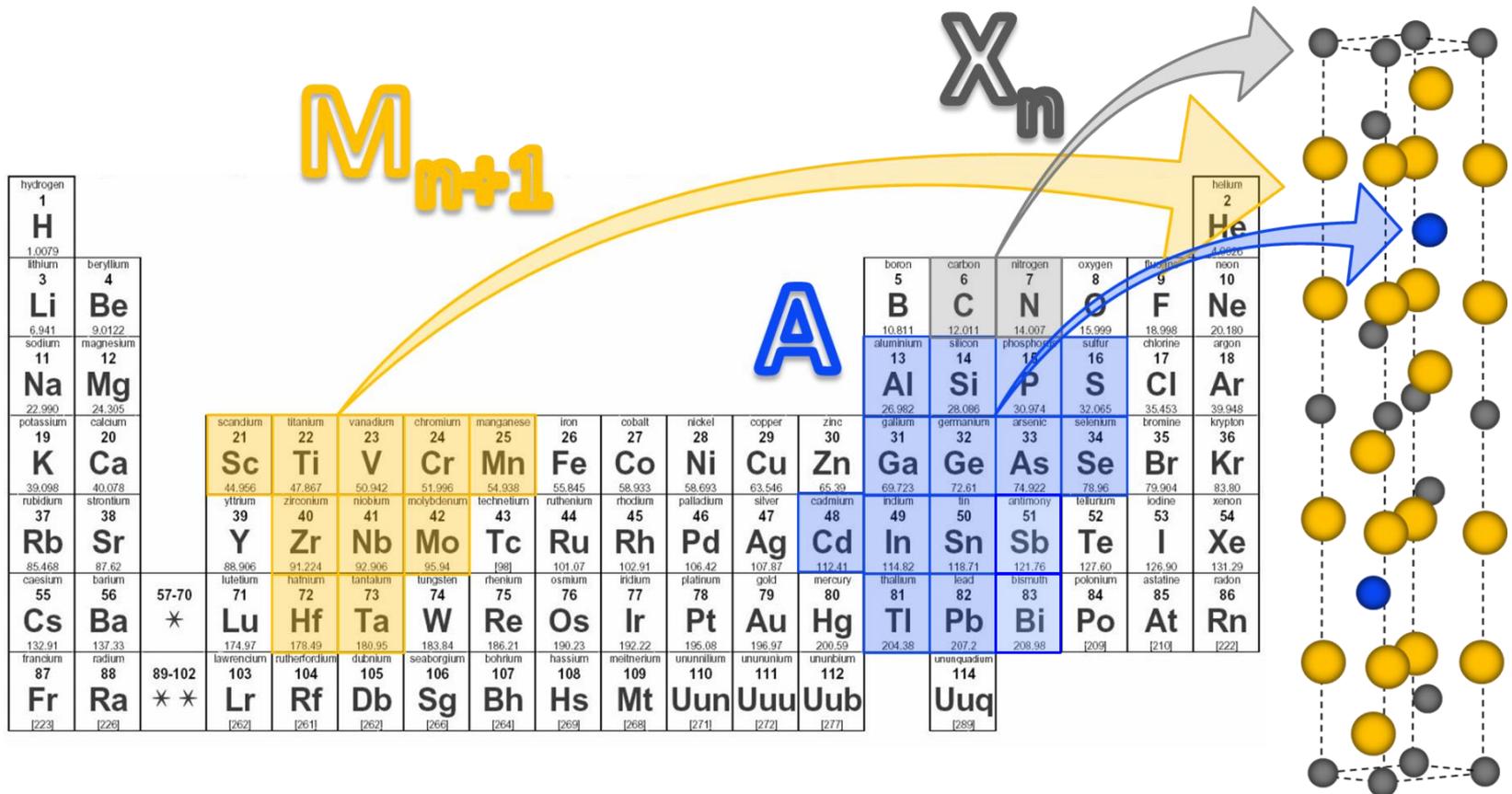
*in collaboration with Sandvik, KIT, MPIE, U Poitiers, Linköping U, Drexel U*



- General Introduction
  - Why MAX Phases?
  - MAX Phases for Advanced Nuclear Systems
- MAX Phases for **Gen-IV LFRs**: **MYRRHA Pump Impeller**
  - **Nb-Al-C System**:  $\text{Nb}_4\text{AlC}_3$
  - **Nb-Zr-Al-C System**:  $(\text{Nb,Zr})_4\text{AlC}_3$
  - Processing towards Mechanical Property Optimization
  - Resistance to Dissolution Corrosion in LBE
  - Resistance to Flow-Assisted Corrosion in LBE
- MAX Phases for **Gen-III+ LWRs**: **Accident-Tolerant Fuels (ATFs)**
  - **Zr-Al-C System**:  $\text{Zr}_3\text{AlC}_2$  &  $\text{Zr}_2\text{AlC}$
- Conclusions

# MAX Phases: Introduction

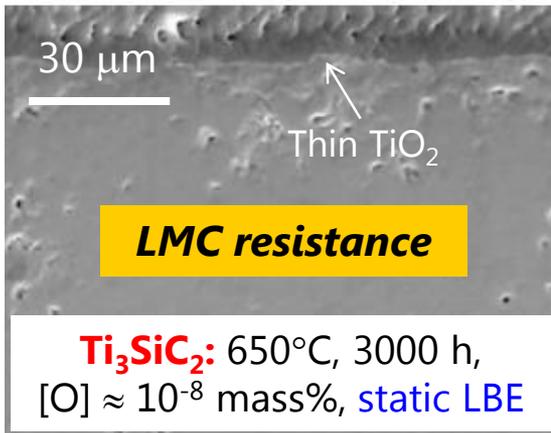
- MAX phases: ternary carbide and nitride compounds described by the general formula  $M_{n+1}AX_n$ , where  $n = 1, 2, 3$
- Laminated crystal structure:  $M_6X$  octahedra interleaved with **A** layers





# MAX Phases: Why?

- Unique combination of physical, chemical, mechanical properties, e.g. LMC resistance, machinability, damage tolerance, irradiation tolerance, ...

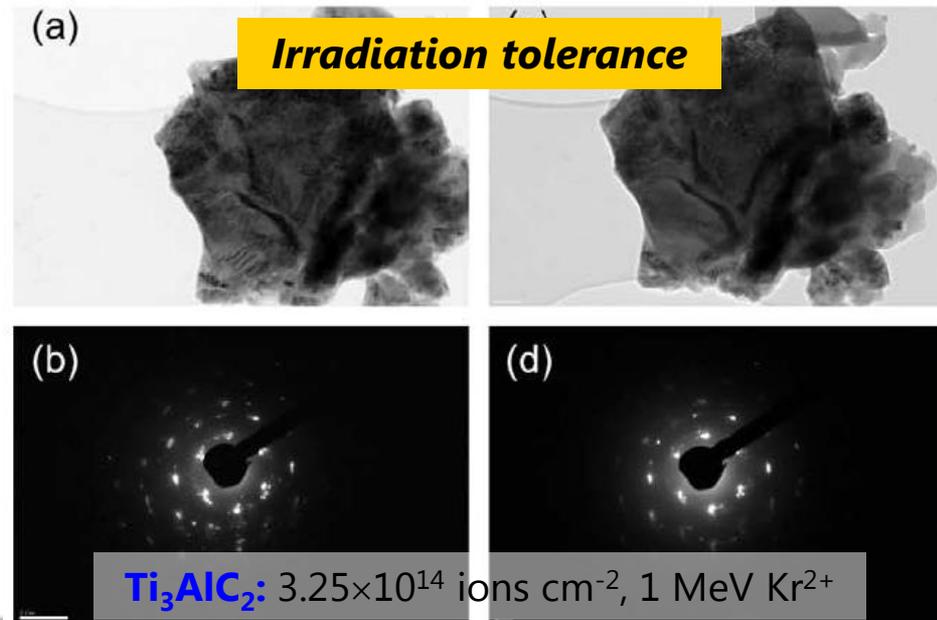


**Ti<sub>3</sub>SiC<sub>2</sub>**: 650°C, 3000 h,  
[O] ≈ 10<sup>-8</sup> mass%, static LBE

(A. Heinzl et al., *Journal of Nuclear Materials* **392** (2009) 255-258)



(M.W. Barsoum & M. Radovic, *Annu. Rev. Mater. Res.* **41** (2011) 195-227)



**Ti<sub>3</sub>AlC<sub>2</sub>**: 3.25×10<sup>14</sup> ions cm<sup>-2</sup>, 1 MeV Kr<sup>2+</sup>

(M.W. Barsoum & M. Radovic, *Annu. Rev. Mater. Res.* **41** (2011) 195-227)

# MAX Phases for Advanced Nuclear Systems

hydrogen 1 H 1.0079																			helium 2 He 4.0026
lithium 3 Li 6.941	beryllium 4 Be 9.0122											boron 5 B 10.811	carbon 6 C 12.011	<del>nitrogen 7 N 14.007</del>	oxygen 8 O 15.999	fluorine 9 F 18.998	neon 10 Ne 20.180		
sodium 11 Na 22.990	magnesium 12 Mg 24.305											aluminium 13 Al 26.982	silicon 14 Si 28.086	<del>phosphorus 15 P 30.974</del>	sulfur 16 S 32.065	chlorine 17 Cl 35.453	argon 18 Ar 39.948		
potassium 19 K 39.098	calcium 20 Ca 40.078	scandium 21 Sc 44.956	titanium 22 Ti 47.887	vanadium 23 V 50.942	chromium 24 Cr 51.996	manganese 25 Mn 54.938	iron 26 Fe 55.845	cobalt 27 Co 58.933	nickel 28 Ni 58.693	copper 29 Cu 63.546	zinc 30 Zn 65.39	gallium 31 Ga 69.723	germanium 32 Ge 72.61	arsenic 33 As 74.922	selenium 34 Se 78.96	bromine 35 Br 79.904	krypton 36 Kr 83.80		
rubidium 37 Rb 85.468	strontium 38 Sr 87.62	yttrium 39 Y 88.906	zirconium 40 Zr 91.224	niobium 41 Nb 92.906	<del>molybdenum 42 Mo 95.94</del>	technetium 43 Tc [98]	ruthenium 44 Ru 101.07	rhodium 45 Rh 102.91	palladium 46 Pd 106.42	silver 47 Ag 107.87	<del>cadmium 48 Cd 112.41</del>	<del>indium 49 In 114.82</del>	tin 50 Sn 118.71	<del>antimony 51 Sb 121.76</del>	tellurium 52 Te 127.60	iodine 53 I 126.90	xenon 54 Xe 131.29		
caesium 55 Cs 132.91	barium 56 Ba 137.33	57-70 *	lutetium 71 Lu 174.97	<del>hafnium 72 Hf 178.49</del>	<del>tantalum 73 Ta 180.95</del>	tungsten 74 W 183.84	rhenium 75 Re 186.21	osmium 76 Os 190.23	iridium 77 Ir 192.22	platinum 78 Pt 195.08	gold 79 Au 196.97	<del>thallium 81 Tl 204.38</del>	<del>lead 82 Pb 207.2</del>	bismuth 83 Bi 208.98	polonium 84 Po [209]	astatine 85 At [210]	radon 86 Rn [222]		
francium 87 Fr [223]	radium 88 Ra [226]	89-102 * *	lawrencium 103 Lr [262]	rutherfordium 104 Rf [261]	dubnium 105 Db [262]	seaborgium 106 Sg [266]	bohrium 107 Bh [264]	hassium 108 Hs [269]	meitnerium 109 Mt [268]	ununnillium 110 Uun [271]	unununium 111 Uuu [272]	ununbium 112 Uub [277]	ununquadium 114 Uuq [289]						

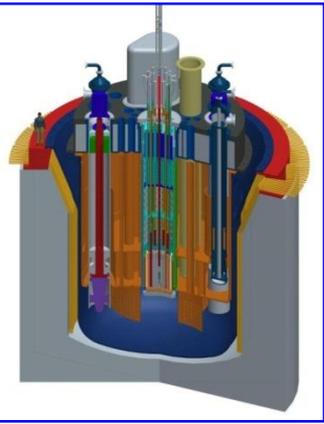
 Large neutron cross-section

 Long-lived isotope <sup>14</sup>C

 Alloying elements in commercial clads (Zircaloy/ZIRLO/15-15Ti SS)

 Systems of interest: Nb-Al-C, Zr-Al-C and solid solutions thereof

- General Introduction
  - Why MAX Phases?
  - MAX Phases for Advanced Nuclear Systems
- MAX Phases for **Gen-IV LFRs: MYRRHA Pump Impeller**
  - **Nb-Al-C System:**  $\text{Nb}_4\text{AlC}_3$
  - **Nb-Zr-Al-C System:**  $(\text{Nb,Zr})_4\text{AlC}_3$
  - Processing towards Mechanical Property Optimization
  - Resistance to Dissolution Corrosion in LBE
  - Resistance to Flow-Assisted Corrosion in LBE
- MAX Phases for **Gen-III+ LWRs: Accident-Tolerant Fuels (ATFs)**
  - **Zr-Al-C System:**  $\text{Zr}_3\text{AlC}_2$  &  $\text{Zr}_2\text{AlC}$
- Conclusions

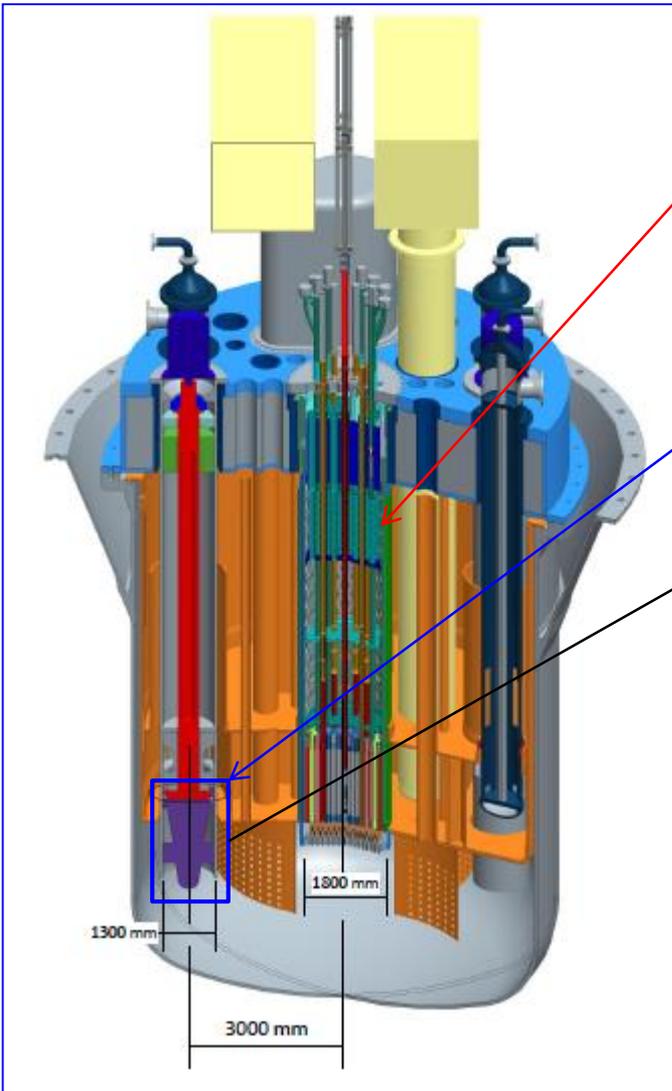


# MAX Phases for Gen-IV LFRs: MYRRHA Pump Impeller



STUDIECENTRUM VOOR KERNENERGIE  
CENTRE D'ETUDE DE L'ENERGIE NUCLEAIRE

# MYRRHA: Pump Impeller Application



Maximum allowable dose at core barrel:  
2 dpa

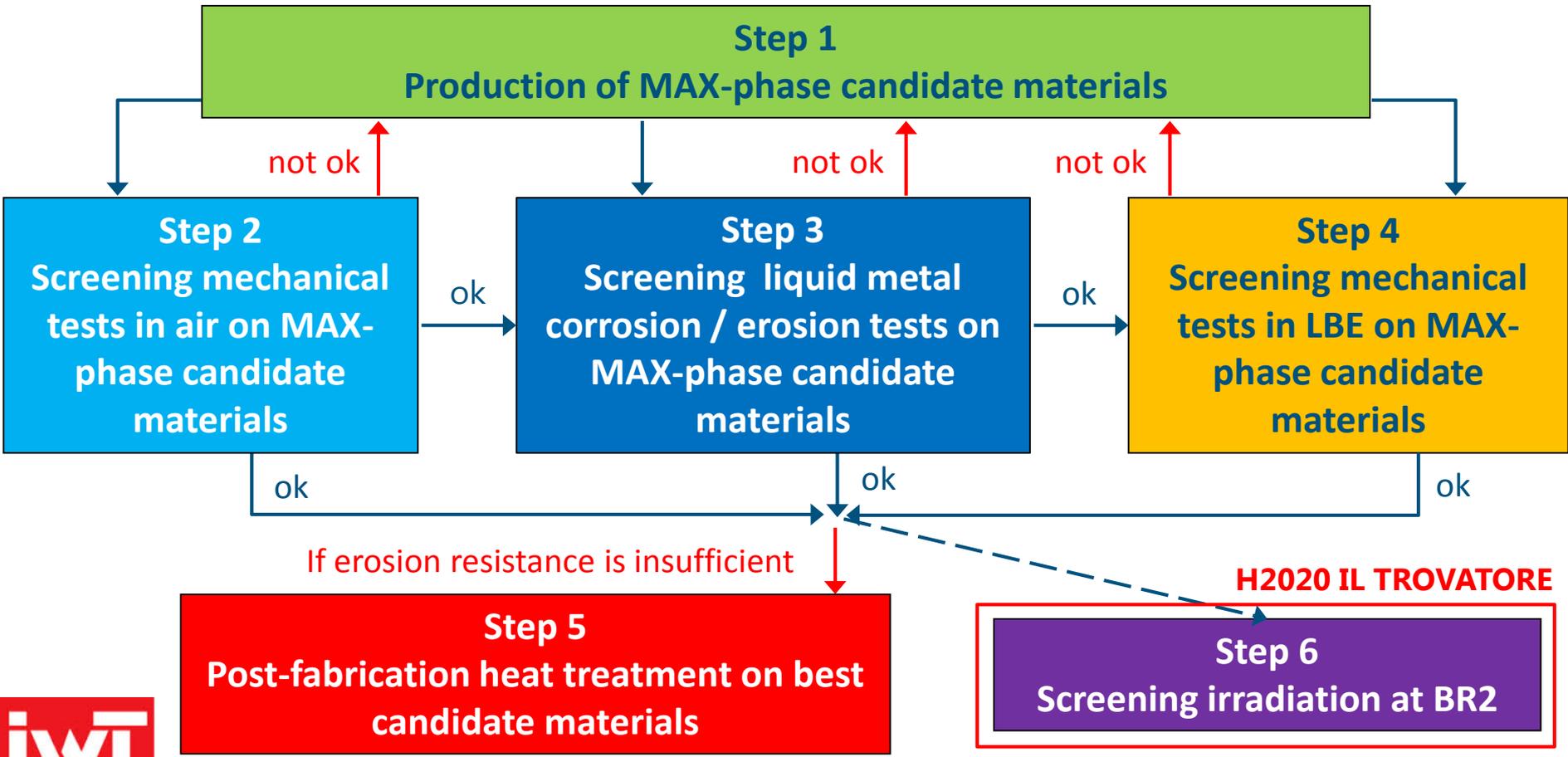
Pump Impeller  
( $< 1$  dpa)

Pump Impeller Rotor\*



## ➤ MYRRHA Pump Impeller Service Conditions:

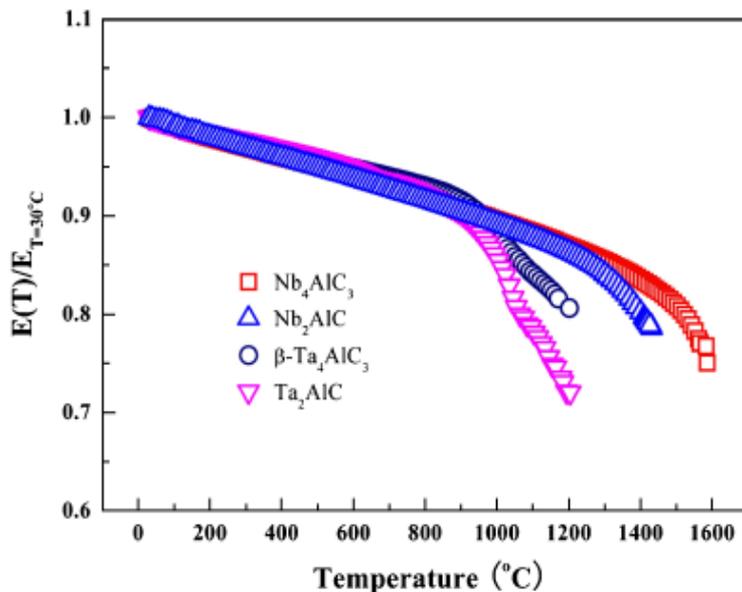
- $T_{\max} \approx 270^{\circ}\text{C}$  ( $T_{\max} \approx 480^{\circ}\text{C}$  in Gen-IV LFRs)
- LBE flow velocity:  $v \approx 10\text{-}20$  m/s locally
- Dissolved oxygen in LBE:  $[\text{O}] \approx 10^{-6}$  mass%
- Fast neutron irradiation dose: low ( $< 1$  dpa)
- Other targeted property requirements (strength, fracture toughness, fatigue resistance)



# Nb-Al-C System: $\text{Nb}_4\text{AlC}_3$

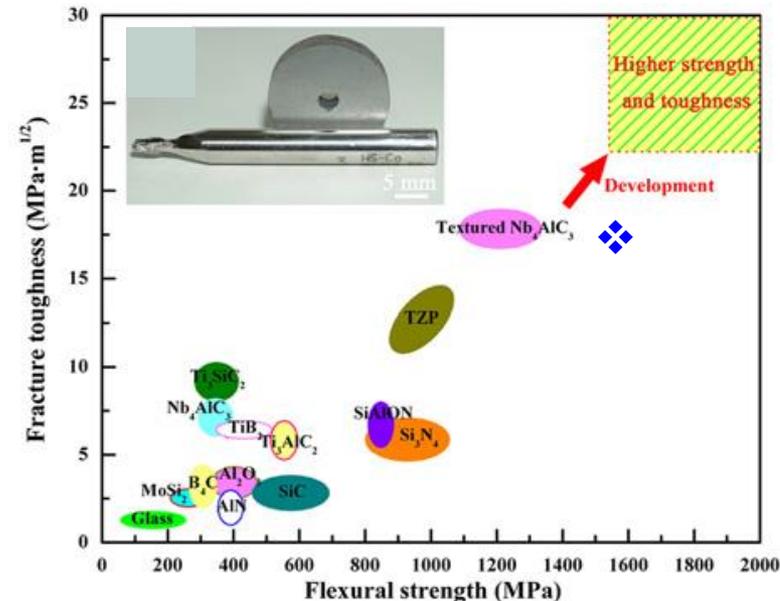
- $\text{Nb}_4\text{AlC}_3$  shows great potential as high-T structural material due to:
  - Good thermal stability as reflected in high-T mechanical properties
  - High fracture toughness ( $7.1 \text{ MPa}\cdot\text{m}^{1/2}$  untextured;  $17.9 \text{ MPa}\cdot\text{m}^{1/2}$  textured)

## Normalized Young's Modulus

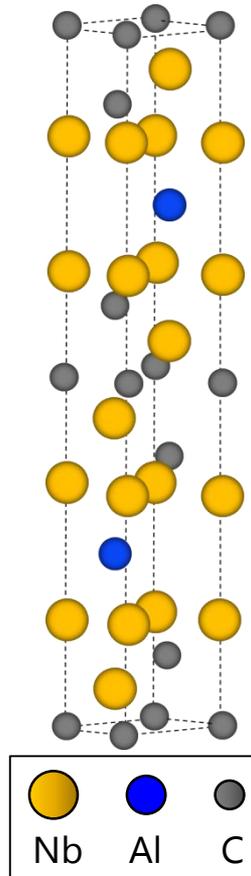


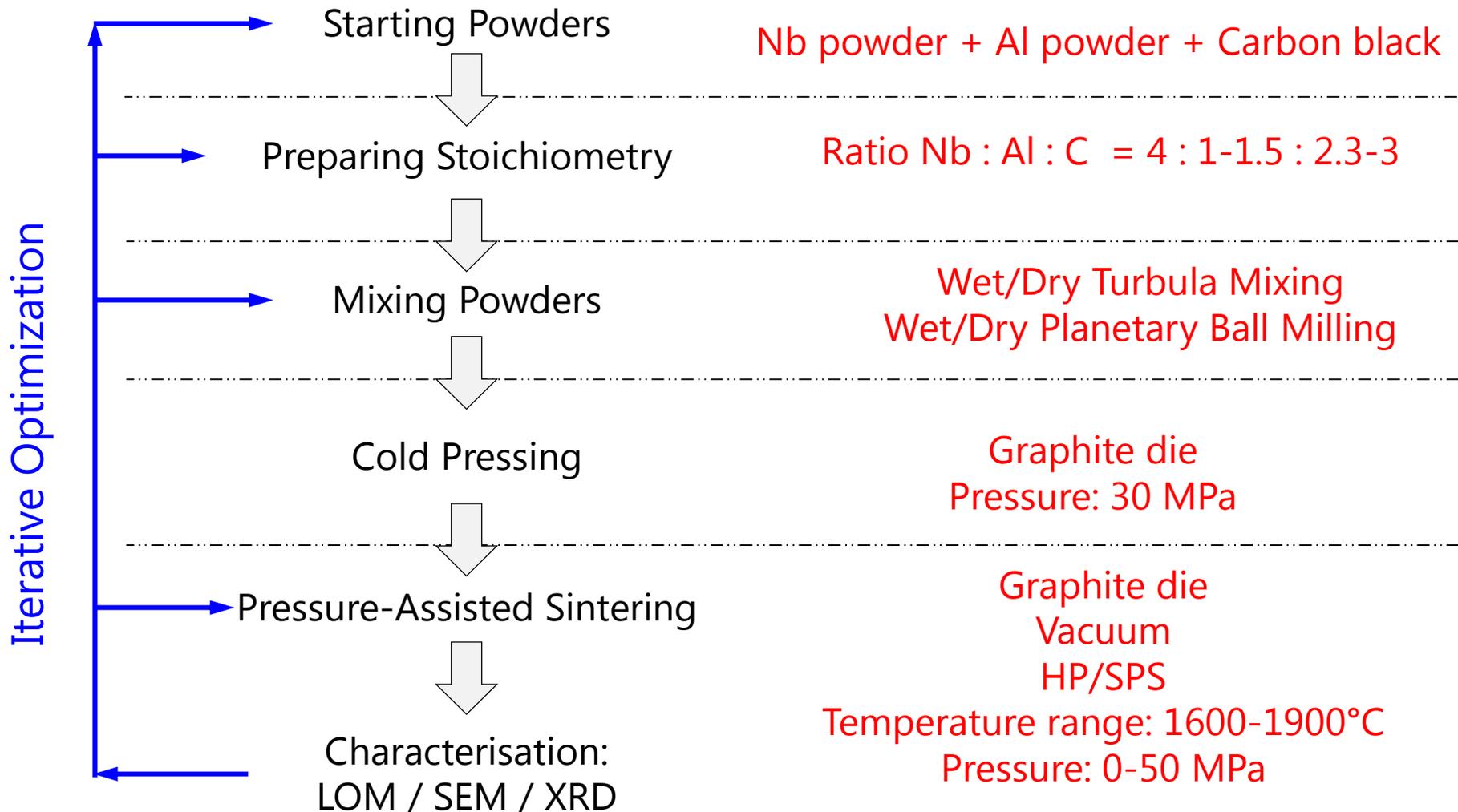
(C.F. Hu et al., *Journal American Ceramic Society* **91** (2008) 2258-2263)

## Fracture Toughness



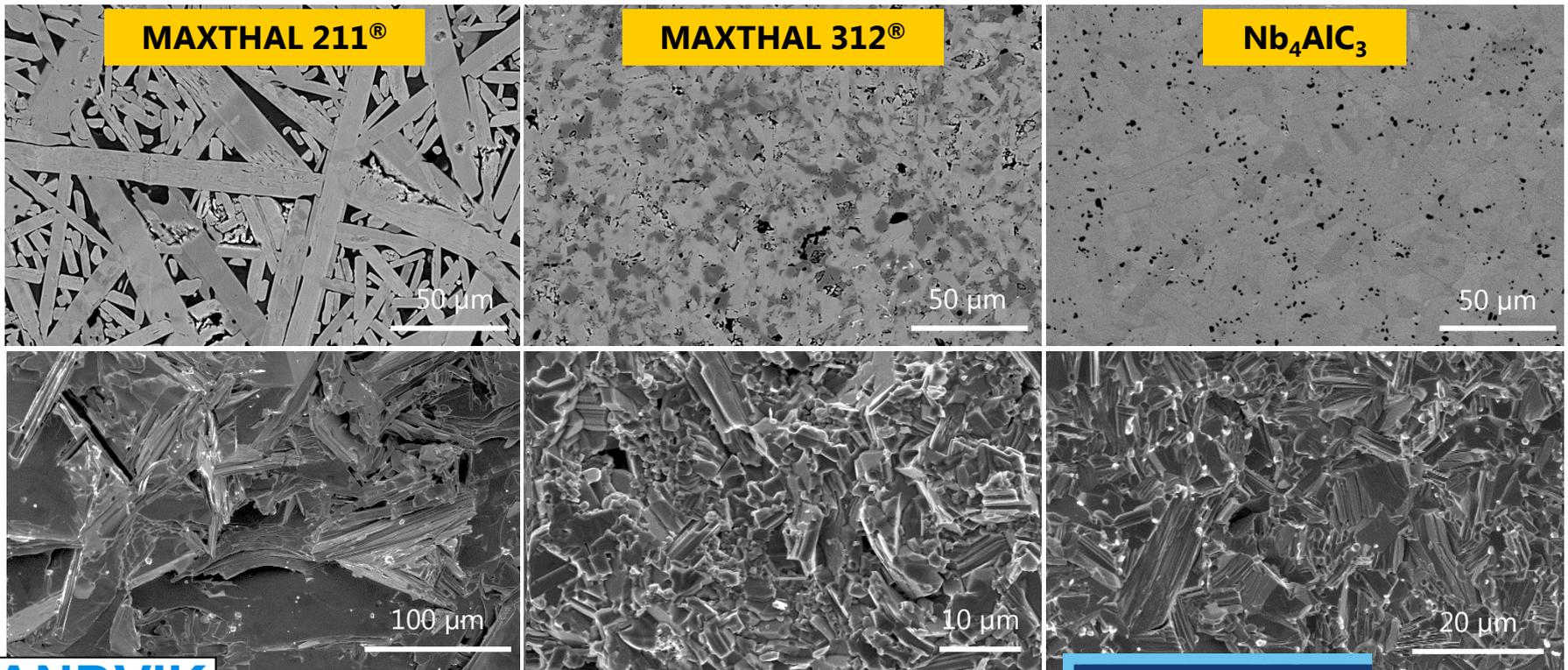
(C.F. Hu et al., *Scripta Materialia* **64** (2011) 765-768)





# Comparison of MAXTHAL 211<sup>®</sup> & 312<sup>®</sup> to Nb<sub>4</sub>AlC<sub>3</sub>

- Commercially-available MAXTHAL 211<sup>®</sup> (nominally Ti<sub>2</sub>AlC) & MAXTHAL 312<sup>®</sup> (nominally Ti<sub>3</sub>SiC<sub>2</sub>) were provided by SANDVIK\*
- Nearly single-phase Nb<sub>4</sub>AlC<sub>3</sub> was produced by SPS (1650°C) at KU Leuven
- All 3 materials were characterized in terms of microstructure & properties



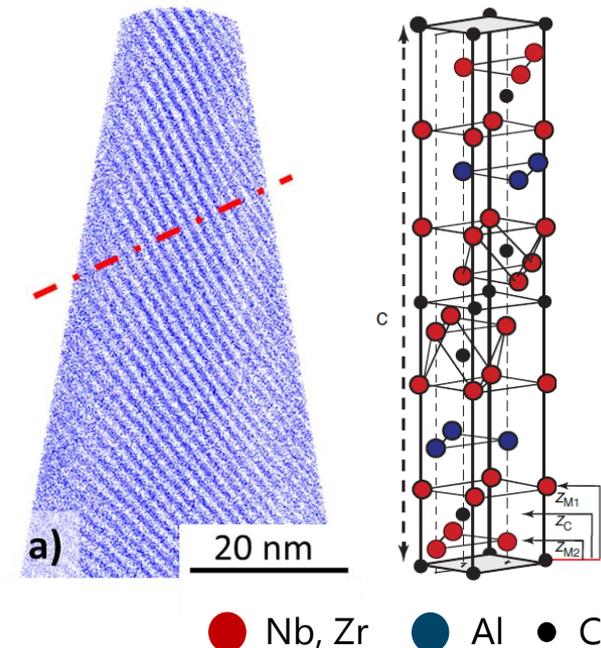
# Comparison of MAXTHAL 211<sup>®</sup> & 312<sup>®</sup> to Nb<sub>4</sub>AlC<sub>3</sub>

Property	MAXTHAL 211 <sup>®</sup>	MAXTHAL 312 <sup>®</sup>	Nb <sub>4</sub> AlC <sub>3</sub>
Composition (wt%)	55 Ti <sub>2</sub> AlC + 38 Ti <sub>3</sub> AlC <sub>2</sub> + 7 TiAl <sub>3</sub>	90 Ti <sub>3</sub> SiC <sub>2</sub> + 10 TiC	98 Nb <sub>4</sub> AlC <sub>3</sub> + 2 Al <sub>2</sub> O <sub>3</sub>
ρ (g/cm <sup>3</sup> )	4.02 (98% ρ <sub>th</sub> )	4.42 (98% ρ <sub>th</sub> )	6.89 (99% ρ <sub>th</sub> )
Grain size (μm)	50-100	5-10	10-20
E (GPa)	265 ± 4	325 ± 5	340 ± 5
G (GPa)	112 ± 2	137 ± 3	137 ± 2
ν	0.17	0.19	0.23
T (°C) where E/E <sub>RT</sub> * = 0.75	± 1250	± 1300	> 1400
H <sub>V,3kg</sub> (GPa)	3.2 ± 0.5	3.8 ± 0.4	3.7 ± 0.3
σ <sub>4PB</sub> (MPa)	210 ± 12	404 ± 44	573 ± 39
K <sub>IC</sub> (MPa·m <sup>1/2</sup> )	6.2 ± 0.2	4.8 ± 0.2	6.6 ± 0.1

\* High-T E measured in vacuum

- Substituting Nb with Zr in  $\text{Nb}_4\text{AlC}_3$  lattice could have beneficial effect on:
  - Smaller neutron cross-section than  $\text{Nb}_4\text{AlC}_3$  ✓
  - Mechanical properties ?
  - Temperature stability ?
- Maximum Zr solid solubility in  $(\text{Nb}_{1-x}\text{Zr}_x)_4\text{AlC}_3$ :
  - $x \approx 18\%$ ; further Zr increase forms  $\text{ZrC}$

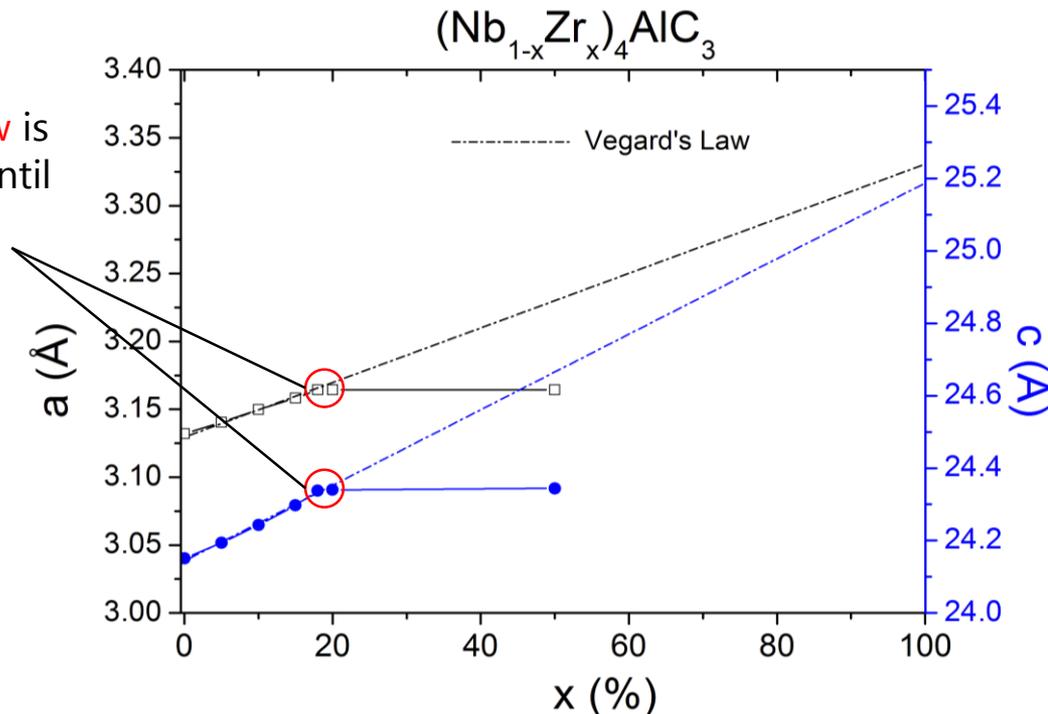
## Atom Probe Tomography



● Nb, Zr   ● Al   ● C



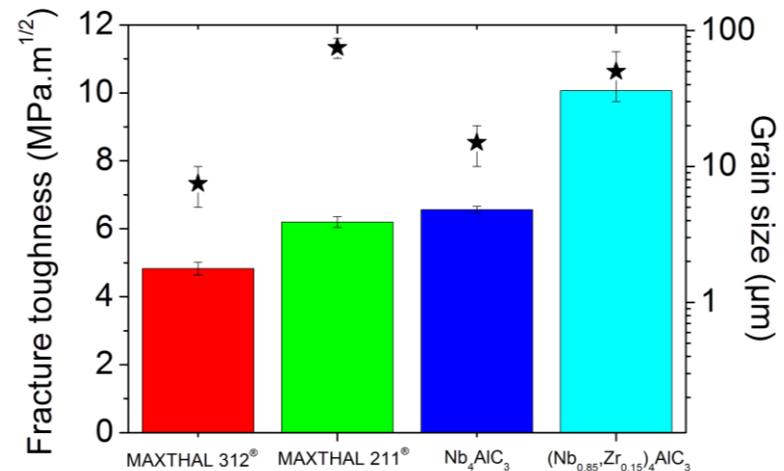
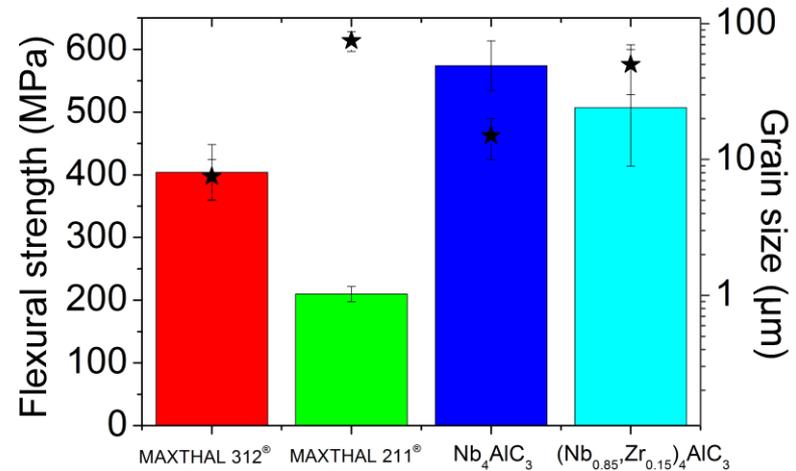
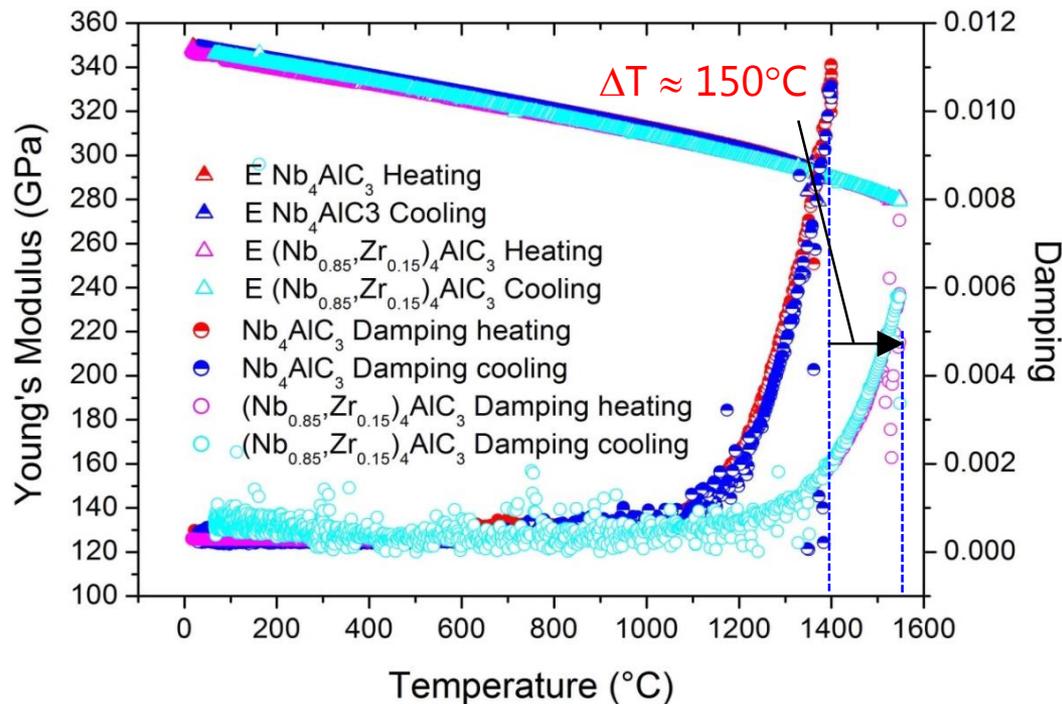
Max-Planck-Institut  
für Eisenforschung GmbH



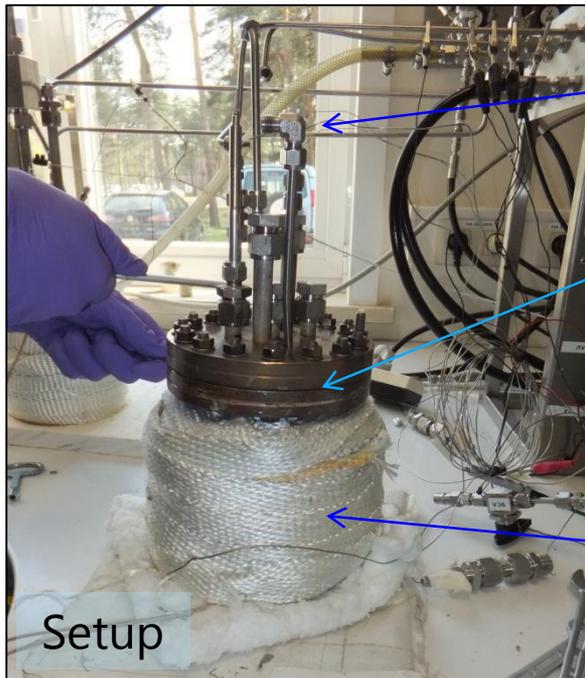
Vegard's law is respected until  $x \approx 18\%$

# Nb-Zr-Al-C System: $\text{Nb}_4\text{AlC}_3$ & $(\text{Nb,Zr})_4\text{AlC}_3$

- Substituting Nb with Zr in  $\text{Nb}_4\text{AlC}_3$  has beneficial effect on:
  - Smaller neutron cross-section than  $\text{Nb}_4\text{AlC}_3$  ✓
  - Mechanical properties ✓
  - Temperature stability ✓



# Resistance to Dissolution Corrosion: Test Setup

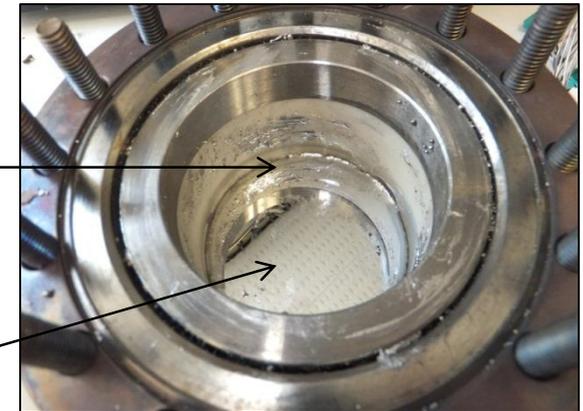


Pipes for gas inlet/outlet

Steel container

Thermal insulation

Setup



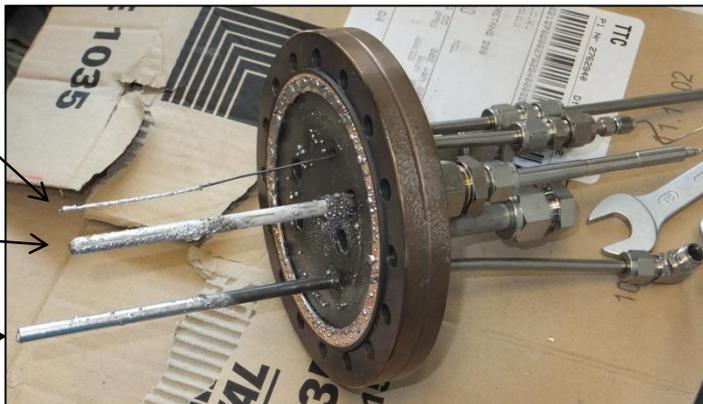
$\text{Al}_2\text{O}_3$  inner liner

Cleaned LBE surface

Thermocouple

Oxygen sensor ( $\text{Bi/Bi}_2\text{O}_3$ )

Gas inlet pipe



Graphite plate

Mo wires

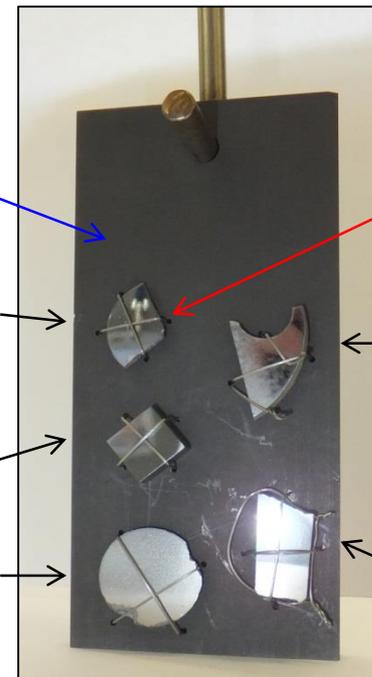
$\text{Nb}_2\text{AlC}$

MAXTHAL 211<sup>®</sup>

$\text{Ti}_3\text{AlC}_2$

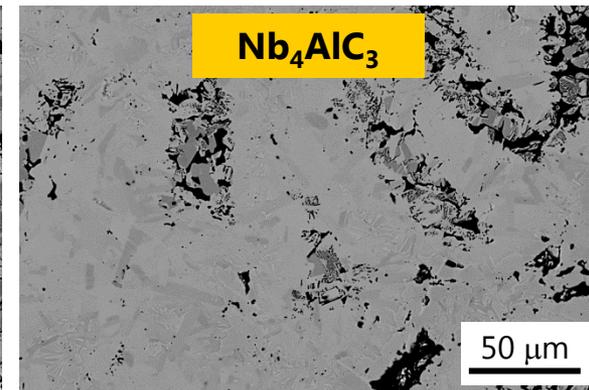
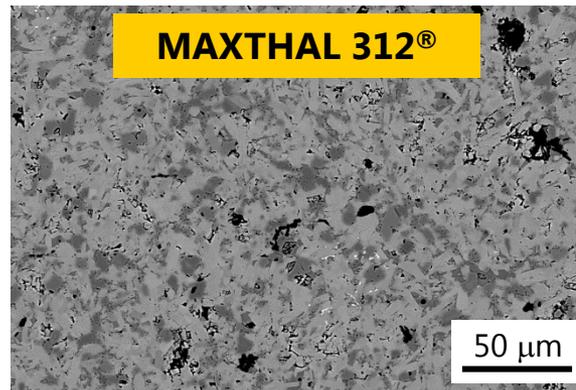
MAXTHAL 211<sup>®</sup> (SANDVIK powder)

MAXTHAL 312<sup>®</sup>

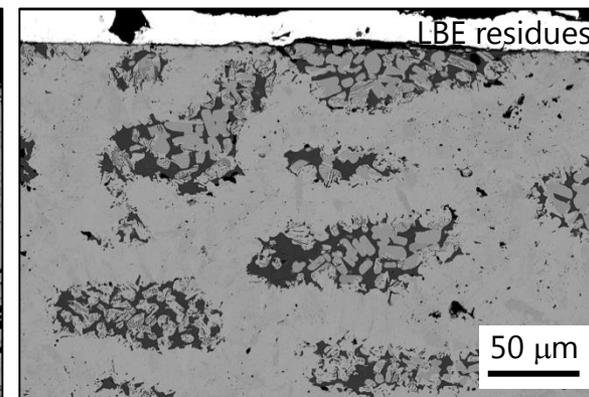
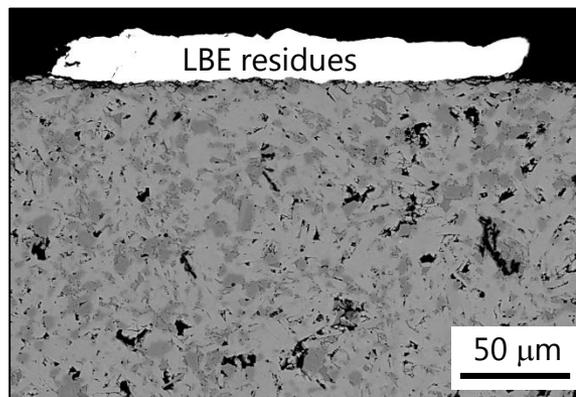
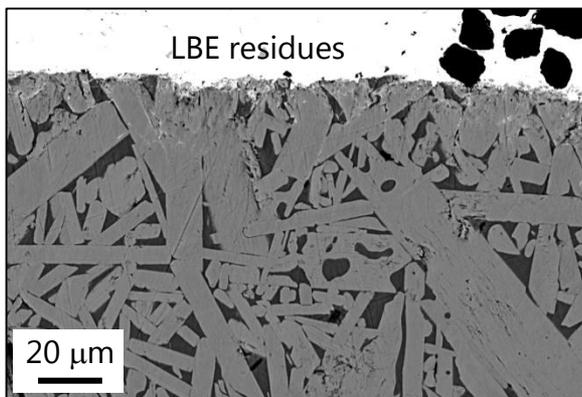


- Materials: MAXTHAL 211® & MAXTHAL 312® provided by SANDVIK; various MAX phases and early-grade Nb<sub>4</sub>AlC<sub>3</sub> produced by SPS (KU Leuven)
- Test: 3501 h, 500°C, [O] ≈ 10<sup>-14</sup> ÷ 10<sup>-10</sup> mass%, static LBE
- **No sign of LBE dissolution attack** was found on the exposed MAX phases

Pristine Materials

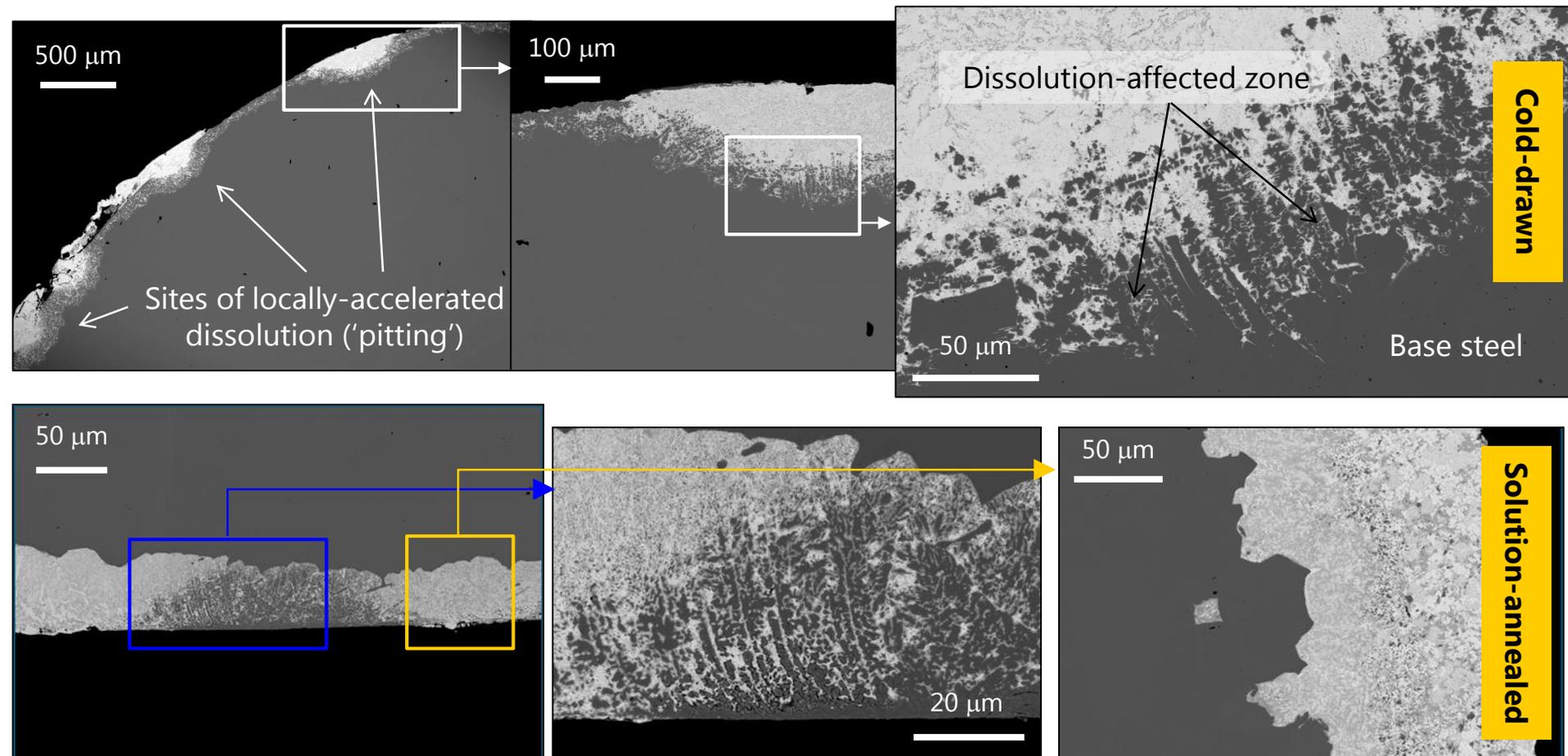


Exposed Materials

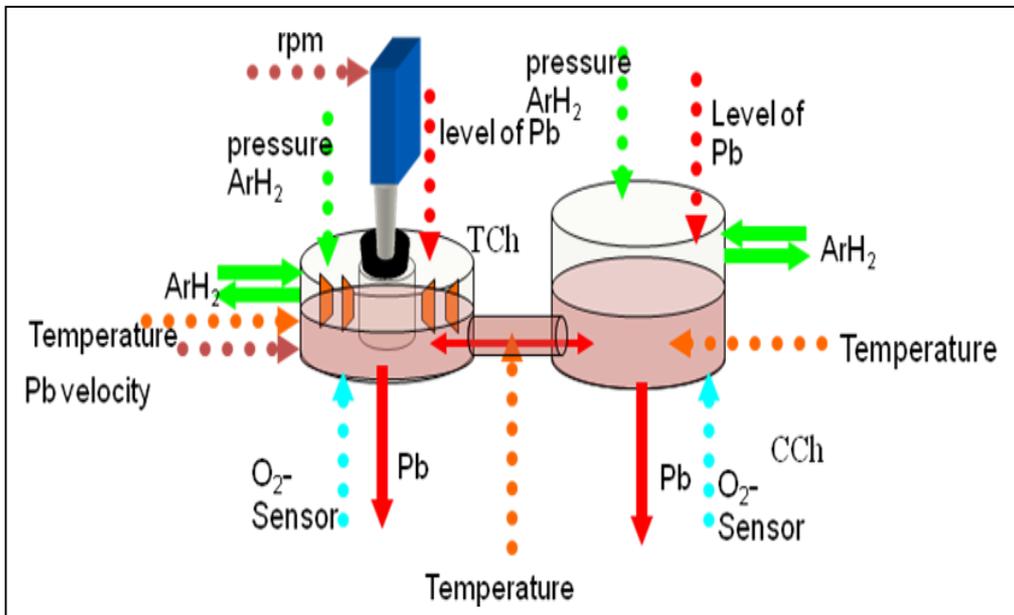


# LMC Behavior of MAX Phases vs. 316L SS

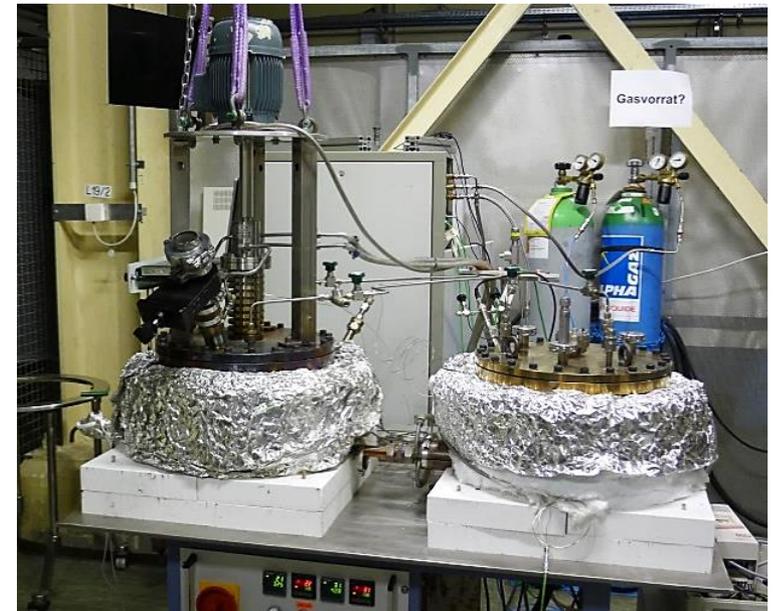
- LMC Test Summary: 3282 h, 500°C,  $[O] \approx 8 \times 10^{-13} \div 3 \times 10^{-8}$  mass%, static LBE
- Dissolution zone: 9-152  $\mu\text{m}$  **solution-annealed** steel; 50-348  $\mu\text{m}$  **cold-drawn** steel



- Test Setup: CORELLA (CORrosion Erosion facility for Liquid Lead Alloy) Facility, KIT, Germany –
  - Material testing in contact with fast-flowing Pb/LBE (>1000 rpm and Pb/LBE flow velocities  $v > 10$  m/s)
  - Monitoring of T and [O] (Bi/Bi<sub>2</sub>O<sub>3</sub> ref. electrode) during testing
  - Testing at [O] constant by conditioning Pb/LBE with appropriate gas mixtures



TCh: test chamber; CCh: conditioning chamber



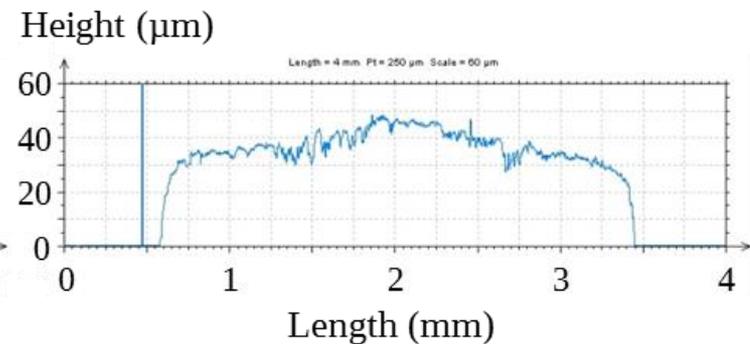
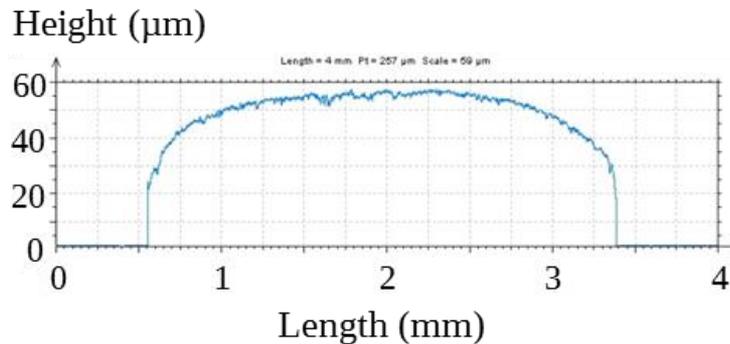
Samples: 60×15×1.5 mm<sup>3</sup>

- White light interferometry was used to assess damages on commercial Maxthal<sup>®</sup> materials: **no signs of flow-assisted corrosion (erosion) damage!**
- Screening tests: 500/500/1000 h, 300°C, liquid LBE, [O]  $\approx 10^{-6}$  mass%,  $v \approx 8$  m/s
- Surface profilometry results on tested MAX materials:

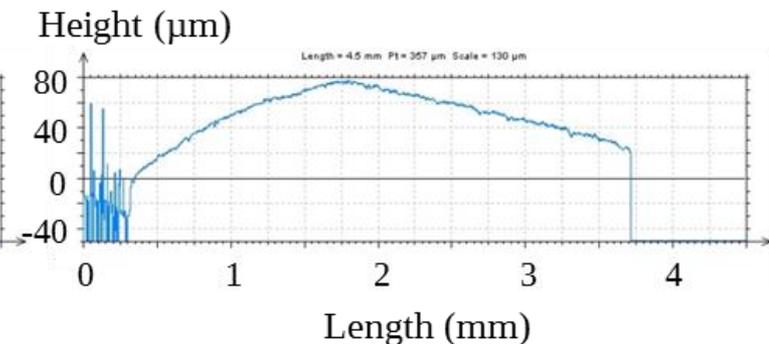
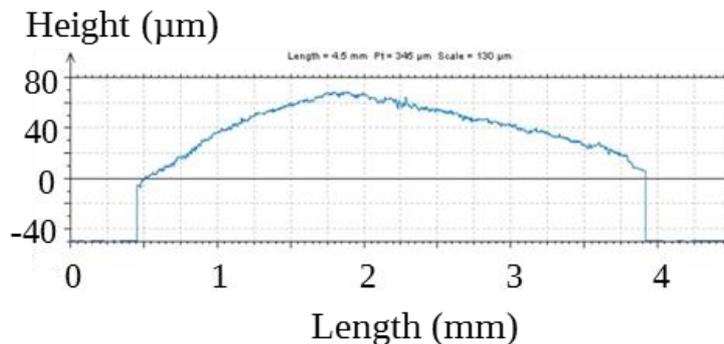
## Pristine Materials

## Exposed Materials

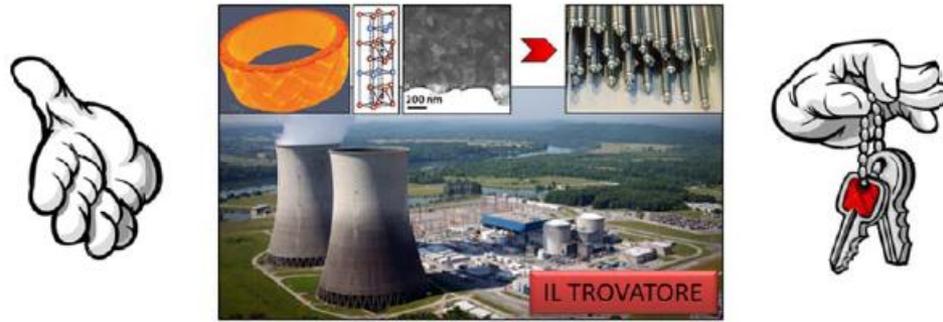
MAXTHAL 211<sup>®</sup>



MAXTHAL 312<sup>®</sup>



- General Introduction
  - Why MAX Phases?
  - MAX Phases for Advanced Nuclear Systems
- MAX Phases for **Gen-IV LFRs**: **MYRRHA Pump Impeller**
  - **Nb-Al-C System**:  $\text{Nb}_4\text{AlC}_3$
  - **Nb-Zr-Al-C System**:  $(\text{Nb,Zr})_4\text{AlC}_3$
  - Processing towards Mechanical Property Optimization
  - Resistance to Dissolution Corrosion in LBE
  - Resistance to Flow-Assisted Corrosion in LBE
- MAX Phases for **Gen-III+ LWRs**: **Accident-Tolerant Fuels (ATFs)**
  - **Zr-Al-C System**:  $\text{Zr}_3\text{AlC}_2$  &  $\text{Zr}_2\text{AlC}$
- Conclusions

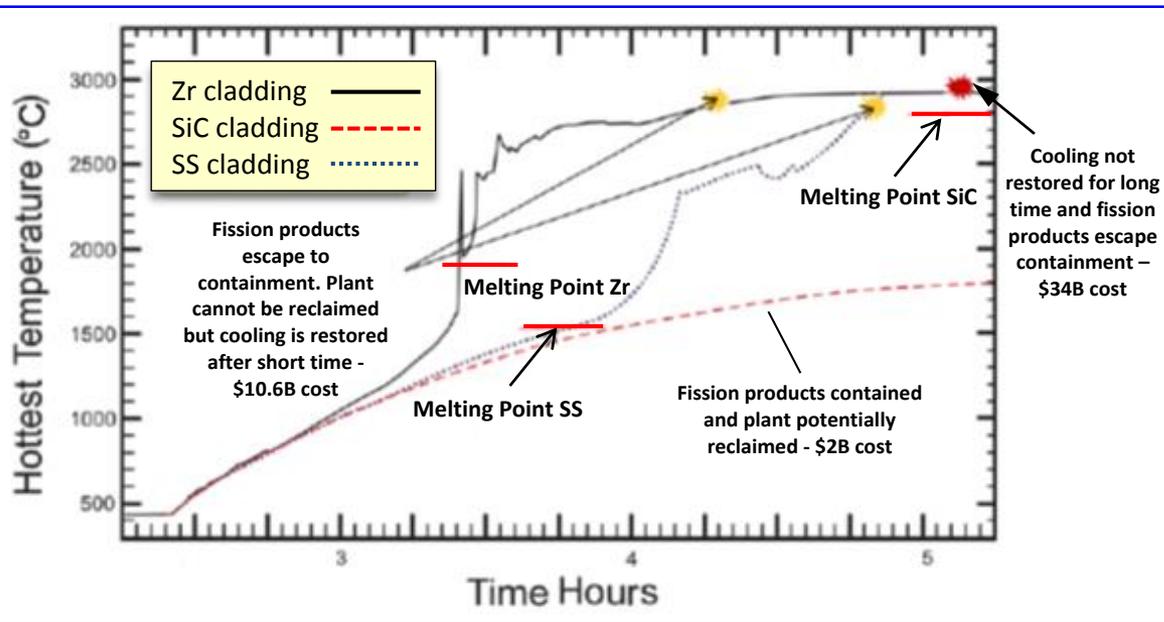


# MAX Phases for Gen-III+ LWRs: Accident-Tolerant Fuels (ATFs)

**KU LEUVEN**



# MAX Phases for ATF Cladding Materials



## Problem Setting:

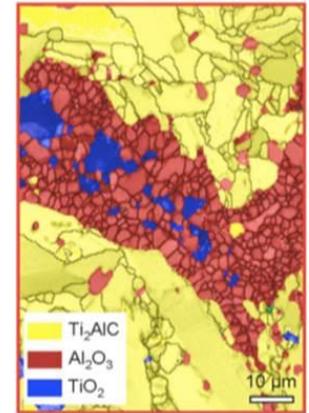
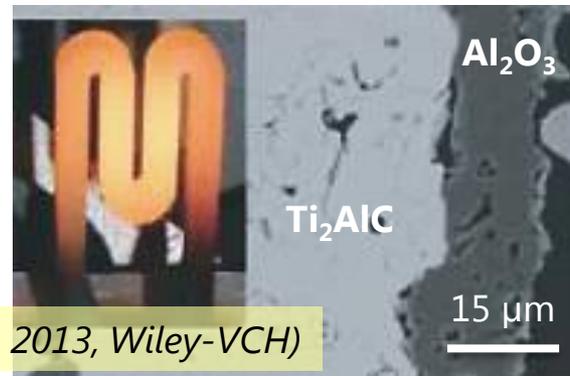
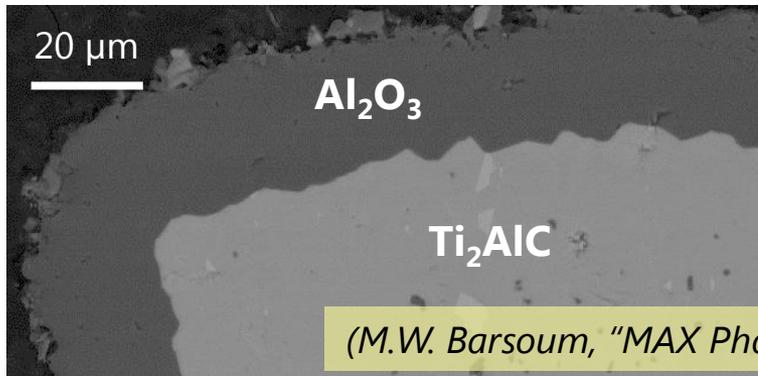
- Loss of coolant & runaway exothermic zircalloy clad/water reactions → cladding failure
- Release of fission products to containment & plant loss
- Release of hydrogen & possible hydrogen explosion
- Escape of radioactive fission products beyond site boundary
- Plant loss & high remediation cost of surrounding area

(E. Lahoda et al., Ref #: RT-TR-14-6, Paper n° 10231, 2014, Westinghouse Electric Company LLC)

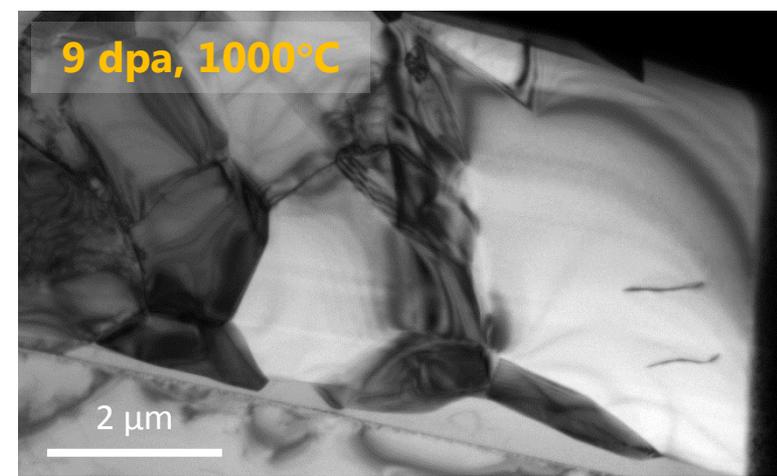
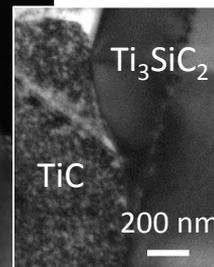
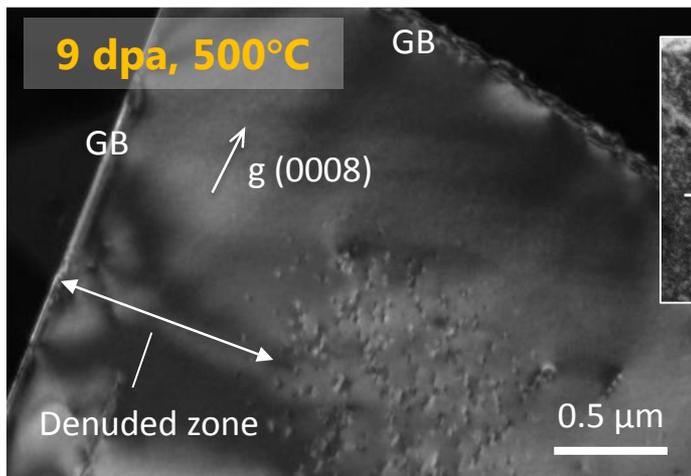
- Use **MAX phases coatings** on **commercial cladding materials** to:
  - Achieve step enhancement in performance of current cladding materials during **nominal operation conditions** & short-lived **design-basis transients (<1200°C)**
  - Maintain hermeticity for prolonged periods (hours to days) under **beyond-design-basis accidents (1200-1700°C)** conditions – **very challenging**

# Potential of MAX Phases for ATF Clads

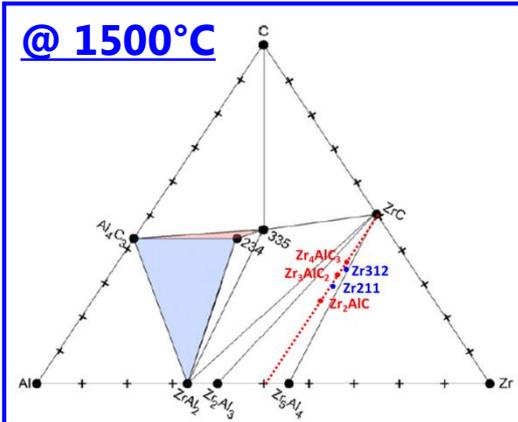
- Certain MAX phases (e.g.  $Ti_2AlC$ ,  $Cr_2AlC$ ) form a **protective oxide scale** ( $Al_2O_3$ ) that does not spall off during **thermal cycling** (e.g.  $Ti_2AlC$ , 8000 cycles to  $1350^\circ C$  in air)
- MAX phases show **self-healing of cracks** in oxidizing environments
- **Self-healing potential for radiation-induced defects!**



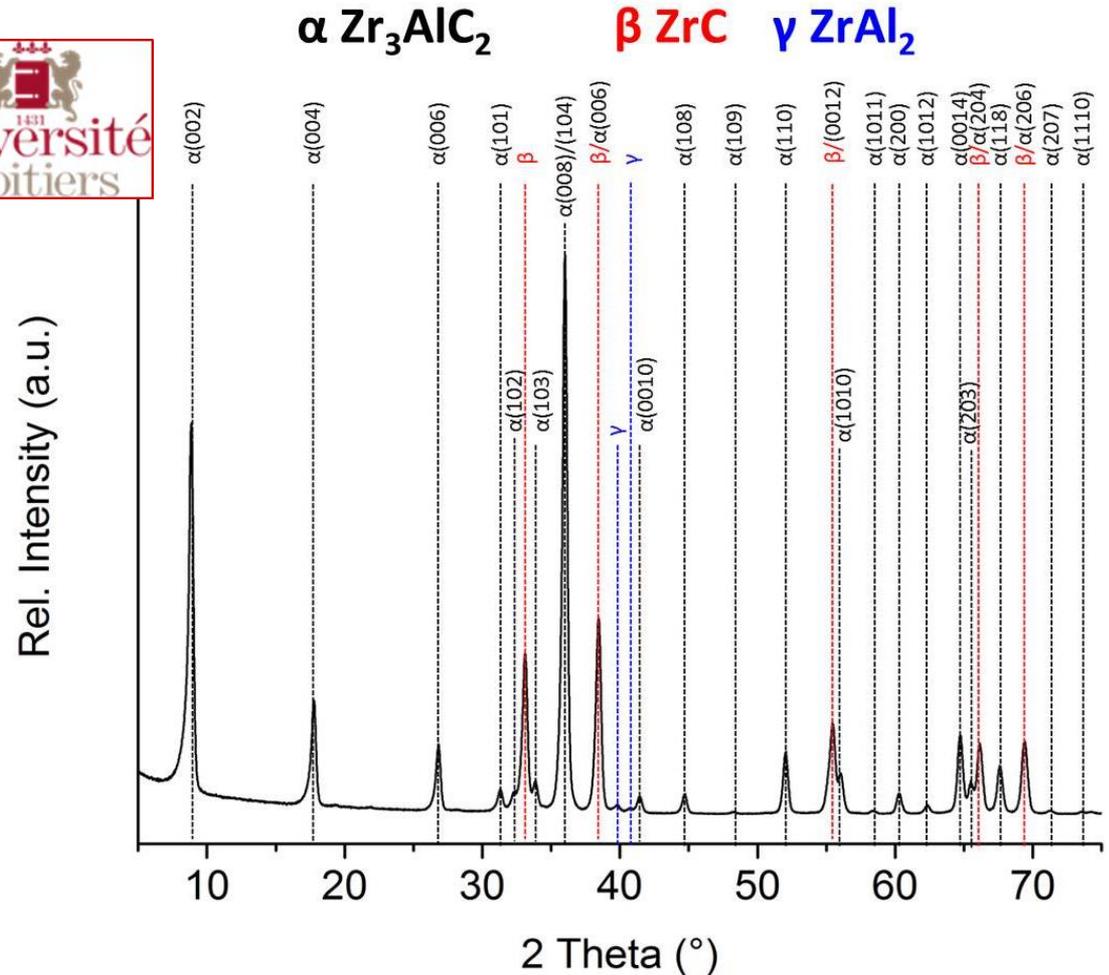
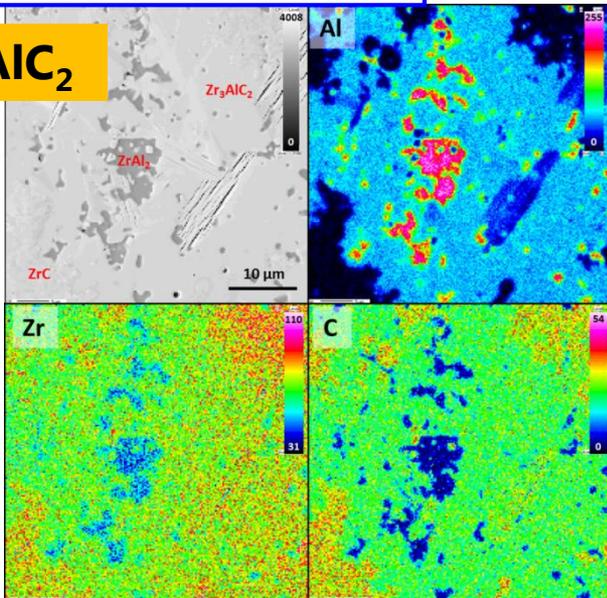
$Ti_3SiC_2$



➤ No MAX phases were synthesized before in the Zr-Al-C system!



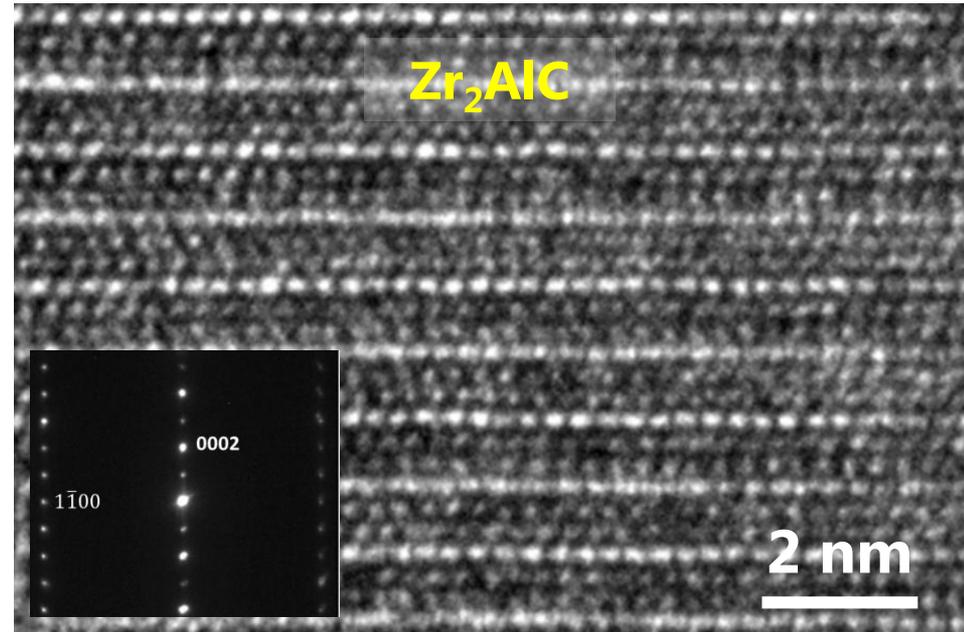
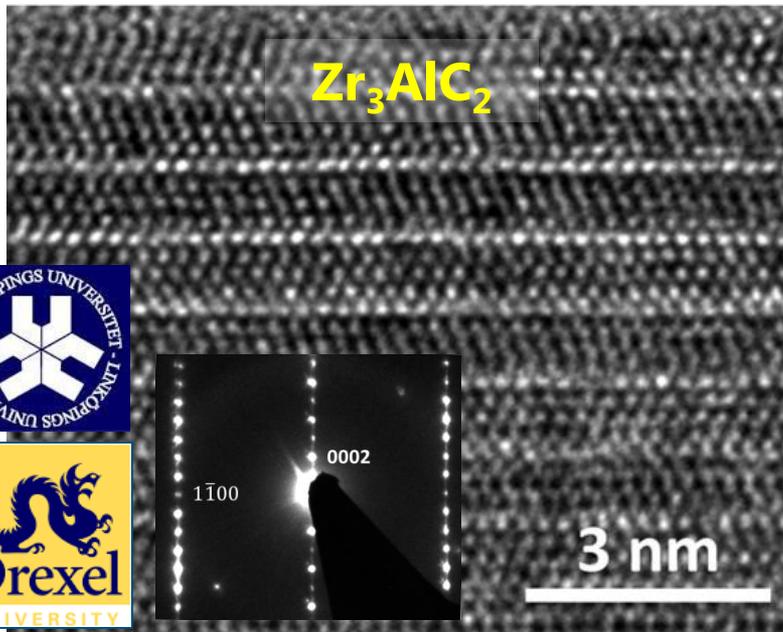
$Zr_3AlC_2$



➤ Lattice parameters determined by XRD and confirmed by HRTEM

$Zr_3AlC_2$	
Space group	$P6_3/mmc$ (194)
Lattice	Hexagonal
a (Å)	3.33308(6)
c (Å)	19.9507(3)

$Zr_2AlC$	
Space group	$P6_3/mmc$ (194)
Lattice	Hexagonal
a (Å)	3.3237(2)
c (Å)	14.5705(4)



- General Introduction
  - Why MAX Phases?
  - MAX Phases for Advanced Nuclear Systems
- MAX Phases for **Gen-IV LFRs**: **MYRRHA Pump Impeller**
  - **Nb-Al-C System**:  $\text{Nb}_4\text{AlC}_3$
  - **Nb-Zr-Al-C System**:  $(\text{Nb,Zr})_4\text{AlC}_3$
  - Processing towards Mechanical Property Optimization
  - Resistance to Dissolution Corrosion in LBE
  - Resistance to Flow-Assisted Corrosion in LBE
- MAX Phases for **Gen-III+ LWRs**: **Accident-Tolerant Fuels (ATFs)**
  - **Zr-Al-C System**:  $\text{Zr}_3\text{AlC}_2$  &  $\text{Zr}_2\text{AlC}$
- **Conclusions**

- Exploring the potential of MAX phases as *candidate materials for specific applications in advanced nuclear systems* entails the following challenges:
  - i. Control of material processing to produce stable phase mixtures that meet the targeted material property requirements (strength, fracture toughness, fatigue resistance) of the end application (e.g. MYRRHA pump impeller)
  - ii. Mechanical testing on the produced materials both in inert/air and heavy liquid metal (Pb/LBE) environments to detect possible environment-assisted material degradation effects (e.g. liquid metal embrittlement)
  - iii. LMC/erosion testing on the produced materials at various test conditions (T, [O], time, flow velocity) so as to understand their behavior
  - iv. Cladding/coolant interaction tests in various coolant states (water, steam) to assess their behavior in service (e.g. ATF cladding materials)
  - v. Screening neutron irradiation campaigns must be designed and performed so as to assess the neutron irradiation tolerance of the produced materials (the criticality of this study depends on the envisaged component, e.g. pump impeller vs. cladding material)



**Thank you for your attention!**  
**Questions?**

**Copyright © 2014 - SCK•CEN**

PLEASE NOTE!

This presentation contains data, information and formats for dedicated use ONLY and may not be copied, distributed or cited without the explicit permission of the SCK•CEN. If this has been obtained, please reference it as a "personal communication. By courtesy of SCK•CEN".

**SCK•CEN**

Studiecentrum voor Kernenergie  
Centre d'Etude de l'Energie Nucléaire  
Belgian Nuclear Research Centre

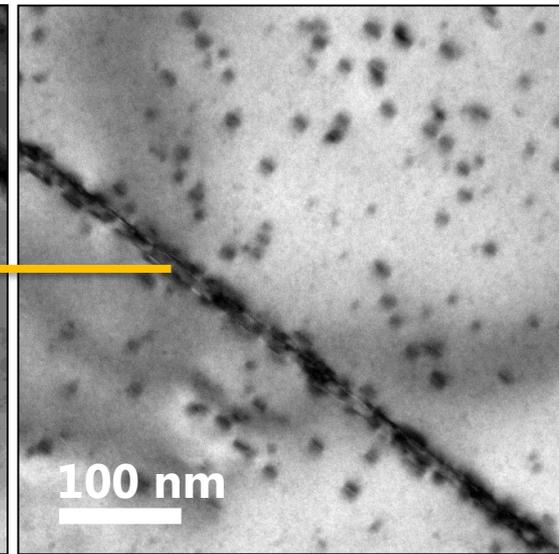
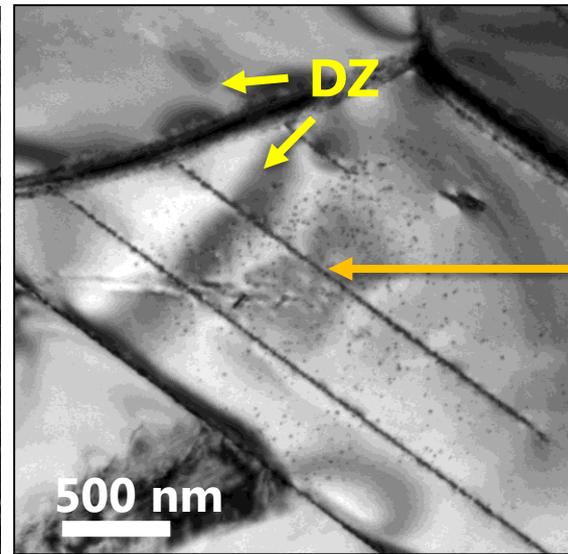
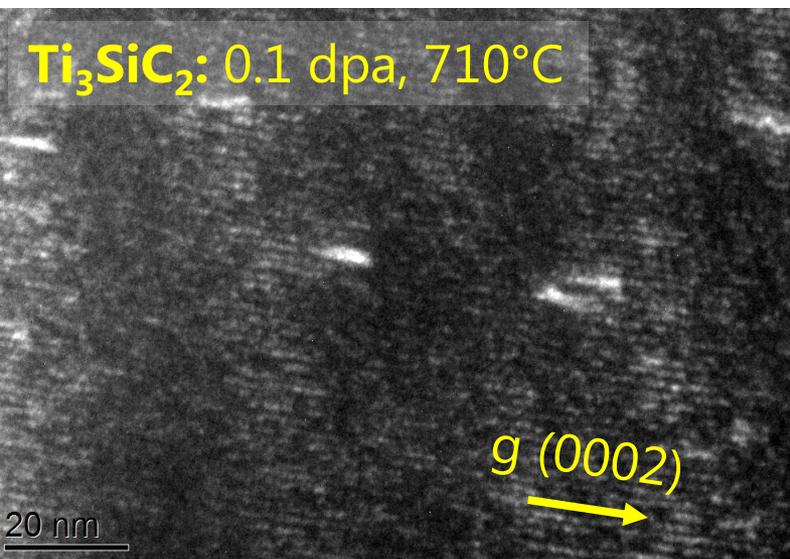
Stichting van Openbaar Nut  
Fondation d'Utilité Publique  
Foundation of Public Utility

Registered Office: Avenue Herrmann-Debrouxlaan 40 – BE-1160 BRUSSELS  
Operational Office: Boeretang 200 – BE-2400 MOL

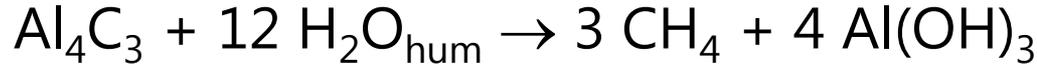


STUDIECENTRUM VOOR KERNENERGIE  
CENTRE D'ETUDE DE L'ENERGIE NUCLEAIRE

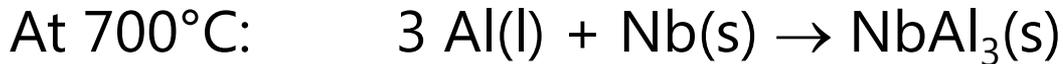
- Investigation of the effect of neutron irradiation on MAX phases:
  - Formation of dislocations loops:  $\mathbf{b} = \frac{1}{2} [0001]$  in the basal plane
  - **Denuded zones (DZ)**: nearly defect-free regions close to grain boundaries
  - Defect clusters are found near stacking faults



- Mixing of Al and C can result in formation of  $\text{Al}_4\text{C}_3$



- Need for fine Nb powder



- Break-up of commercial, coarse Nb (~300 μm)

- Ball-milling: resulted in Nb flakes
- Hydrogenating + Ball Milling

